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(12) UK Patent Application (19) GB (11) 2 083 499 A

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(21) Application No 8028735

(22) Date of filing 5 Sep 1980

(43) Application published  
24 Mar 1982

(51) INT CL<sup>3</sup>  
C22C 38/50

(52) Domestic classification

C7A 782 A23Y A249

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A25Y A28X A28Y A30Y

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(58) Field of search

C7A

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(54) Austenic steel

(57) An austenitic alloy of iron,  
chromium, nickel and aluminium  
contains

15 to 25% chromium

15 to 20% nickel

2 to 4% aluminium

0.02 to 0.15% carbon

0.1 to 0.8% titanium

0.2 to 2.0% silicon

0.2 to 2.0% manganese

and optionally from 0 to 0.8% of  
yttrium or hafnium or any other active  
rare earth metal, or of zirconium or of  
the mixture of rare earth metals  
known as mischmetall, the balance  
being iron and incidental amounts of  
other alloying elements.

The alloy provides resistance to  
oxidation and sulphidation together  
with weldability and hot strength.

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# SPECIFICATION Austenitic steel

The present invention relates to austenitic alloys of iron, chromium and nickel, containing additions of aluminium and titanium, with or without additions of either hafnium or yttrium.

5 The present invention provides an austenitic alloy of iron, chromium, nickel and aluminium containing 5

15 to 25% chromium

15 to 20% nickel

2 to 4% aluminium

10 0.02 to 0.15% carbon 10

0.1 to 0.8% titanium

0.2 to 2.0% silicon

0.2 to 2.0% manganese

and optionally from 0 to 0.8% of yttrium or hafnium or any other active rare earth metal, or of zirconium 15  
or of the mixture of rare earth metals known as mischmetall, the balance being iron and incidental 15  
amounts of other alloying elements. In our copending applications Nos. 8006738 and 8006739 we  
disclosed ferritic alloys of iron, chromium and aluminium containing titanium and hafnium respectively  
and providing resistance to oxidation. It has been found that the ferritic alloys described in our  
copending applications, although possessing adequate oxidation and/or sulphidation resistance, have 20  
insufficient hot strength for use in some environments or fail to find application due to their inherently 20  
poor welding and fabricating characteristics.

Those skilled in the art will appreciate that a sufficient amount of chromium must be present to  
provide a basic oxidation resistance. On the other hand, since a low nickel content is desirable from the  
point of view of sulphidation resistance, to maintain an austenitic structure the chromium content should  
25 be kept to the minimum consistent with adequate oxidation resistance. The additional benefit 25  
contributed by small but significant additions of either hafnium or yttrium does in some instances  
appear worthwhile. The provision of titanium stabilises the alloy against intercrystalline corrosion  
effects. A high silicon content, whilst possibly being beneficial as regards oxidation resistance, is to be  
avoided on account of its possible embrittlement effect. Similarly, a sufficiently high manganese content  
30 to be effective as a delta ferrite suppressor is not advisable on account of its detracting from oxidation 30  
resistance.

Optionally up to 0.8%, e.g. 0.05 to 0.80%, of either yttrium or hafnium or any other of the active  
rare earth metals may be added, or of zirconium, or of the mixture of rare earth metals known as  
mischmetall. The balance of the composition is iron and incidental amounts of other alloying elements  
35 not specifically added. The presence of incidental amounts of copper, cobalt, molybdenum and tungsten 35  
above the impurity level may be tolerated provided they are not present in excess. Other elements such  
as sulphur and phosphorus may be present but these are impurities which are not desirable.

In all cases, however, the composition balance is such as to ensure a virtually non-magnetic alloy.

The alloys may be manufactured by processes normally used for making alloys of this general  
40 type. For instance, they may be made by induction melting, either in air or using inert atmosphere or 40  
vacuum as appropriate, cast into ingots and subsequently forged or rolled into billet or slab prior to  
working down to plate, sheet, strip, bar wire, tube or any other commercially saleable form.

In a typical small scale process for producing an austenitic steel of this invention, a charge of high  
purity iron, nickel pellet and low carbon ferrochromium is melted down in a basic lined induction  
45 furnace, either in air under a base slag, or under an inert atmosphere or in vacuo, without slag, as 45  
appropriate. When completely melted, the appropriate additions of aluminium, ferrotitanium (and  
hafnium or yttrium or other special metal addition) are added, in that order, the metal brought to  
temperature and cast into an appropriate ingot mould.

The invention will be illustrated by the following examples. Alloys according to the invention were  
50 prepared having the compositions given below by the process described above. The size of the melts 50  
was 10 kg each, giving a  $2\frac{1}{2}$ " (60 mm) square ingot which was heated to 1150°C and forged under a  
10 cwt. hammer to produce suitable test bar.

Example	C	Si	Mn	Cr	Ni	Al	Ti	Y	Hf
A	0.077	0.73	0.95	17.64	17.40	2.48	.41	—	—
B	0.078	0.72	0.95	17.68	17.48	2.48	.39	.40	—
C	0.064	0.74	0.94	17.52	17.48	2.70	.41	—	.52

5 The resistance of these steels to oxidation was compared by the following test procedure:  
Specimens some  $\frac{1}{2}$ " (13 mm) in diameter by  $1\frac{1}{2}$ " (30 mm) long were machined from bar and ground to a 120 grit finish. They were washed and cleaned in alcohol prior to test.

The test was of relatively short duration but involved cycling between ambient and test temperature. The test chamber was an alumina tube 2" (50 mm) internal diameter in which the sample was positioned across an open ended alumina boat. Heating was by means of the concentric electric furnace, the temperature being measured by reference to a noble metal thermocouple, the hot junction of which was immediately above the specimen. The test atmosphere was produced by burning natural gas using excess air over that required for combustion, the flow rates being 1.4 cubic foot and 14 cubic foot (0.04 and 0.4 cubic metres) per hour respectively for gas and air. The combustion product, a mixture of nitrogen, oxygen, carbon dioxide and steam, was preheated to test temperature before passing through the test chamber; test temperature was established prior to inserting the sample so that heating was rapid. Each test cycle was for six hours; after this the specimens were removed from the test chamber and cooled in a closed container so that any scale which became detached was collected. When cold, the specimen was weighed, together with any detached scale and then scrubbed with a stiff bristle brush to remove any loosely adherent oxide prior to reweighing to obtain the starting weight for the next cycle. The whole procedure was repeated for a total of seven cycles and the total gain in weight, that is the sum of the individual gains, expressed as milligrams per square centimetre for the 42-hour period, using the original surface area for the untested specimen, was taken as the scaling index.

25 The scaling indices for the steels tested were as follows:

	Example A	Example B	Example C	Typical AISI 310
1100°C	7.7	2.3	4.3	6
1150°C	52.3	30.5	34.2	11
1200°C	119.3	66.6	103.1	17
1250°C	133.0	141.1	146.9	30

Thus for service at temperatures up to 1100°C in oxidising atmospheres the alloys covered by the present application are at least equivalent to the standard material known as AISI 310. None of the materials were suitable for use at higher temperatures.

35 To make a rough assessment of resistance to sulphidation, similar test specimens to those used above were prepared and weighed and then half immersed in an intimately ground mixture of 90% sodium sulphate and 10% sodium chloride contained in alumina boats. These were then placed on a stainless steel tray, with samples of a number of other different materials similarly treated, and the whole placed in a muffle furnace and heated to 900°C for six hours. The tray was then withdrawn, the samples allowed to cool, cleaned as far as possible in hot water and then cathodically descaled in a bath of molten sodium hydroxide at 350°C, using a current of 9 amperes for 20 minutes. After descaling and thorough washing, they were dried in alcohol and reweighed. Losses in weight, expressed in mgm. per sq. cm., were as follows:—

Example	Weight Loss
A	4.4
B	1.3
C	2.2

These figures were as low as those found with any other steel tested, including the special iron-

chromium-aluminium alloys of our copending applications referred to above. In general, alloys with high nickel contents but not containing aluminium failed catastrophically, examples being:

	Sample	Weight Loss	
5	A.I.S.I. 310 (25 Cr—20 Ni)	1114	5
	Ni—Cr Alloy (80Ni—20 Cr)	1447	

To check the relative hot strengths of the proposed materials, hot tensile tests were carried out at a number of temperatures, with a strain rate of three inches per minute. These indicated the hot strength to be similar to that of A.I.S.I. 310 and considerably higher than that of the iron-chromium-aluminium alloys. 10

Laboratory scale welding trials, with deposits laid down on the edges of 1" x  $\frac{1}{4}$ " coupons, indicated freedom from cracking within the parent metal.

The alloys exemplified above can be seen to provide a weldable, relatively strong series of alloys, similar in general characteristics to the well known A.I.S.I. 310 heat resisting material but having the advantage of resistance to sulphidation attack exhibited by the iron-chromium-aluminium alloys. Alternatively, they exhibit the resistance to sulphidation attack of the iron-chromium-aluminium alloys but have the advantage over these of higher hot strength and weldability, making them suitable for tube manufacture for use in difficult environments without the need for cladding and rendering them capable of being used in welded fabrications, their only disadvantage being a lower resistance to pure oxidation as compared with the more advanced iron-chromium-aluminium alloys. These latter properties, however, can only adequately be realised with adequate support in applications such as electric resistance elements. 15 20

#### CLAIMS

- 25 1. An austenitic alloy of iron, chromium, nickel and aluminium containing 25 15 to 25% chromium 15 to 20% nickel 2 to 4% aluminium 0.02 to 0.15% carbon
- 30 0.1 to 0.8% titanium 30 0.2 to 2.0% silicon 0.2 to 2.0% manganese

and optionally from 0 to 0.8% of yttrium or hafnium or any other active rare earth metal, or of zirconium or of the mixture of rare earth metals known as mischmetal, the balance being iron and incidental amounts of other alloying elements. 35

2. An alloy as in claim 1 containing from 17—20% chromium.
3. An alloy as in claim 1 or 2 containing 17—20% nickel.
4. An alloy as in any preceding claim containing titanium such that the content is not less than 4 times the carbon content but not more than 0.8%.
- 40 5. An alloy as claimed in any preceding claim containing about 2.5% aluminium. 40 6. An alloy as claimed in any preceding claim containing 0.05 to 0.8% yttrium. 7. An alloy as in claims 1 to 5 containing 0.05 to 0.8% hafnium. 8. An alloy as in claims 1 to 5 containing 0.05 to 0.8% zirconium. 9. An alloy as in claims 1 to 5 containing 0.05 to 0.8% of a rare earth metal or of the mixture of 45 rare earth metals known as mischmetal. 45 10. An alloy substantially as hereinbefore described in any one of Examples A to C. 11. Bar, billet, wire, slab, plate, sheet, tube or forgings in an alloy as claimed in any one of the preceding claims.